CH106BS/CH206BS: ENGINEERING CHEMISTRY LAB

B.Tech. I Year II Sem.

L T P C 0 0 3 1.5

Course Objectives: The course consists of experiments related to the principles of chemistry required for engineering student. The student will learn:

- Estimation of hardness and chloride content in water to check its suitability for drinking purpose.
- To determine the rate constant of reactions from concentrations as an function of time.
- The measurement of physical properties like adsorption and viscosity.
- To synthesize the drug molecules and check the purity of organic molecules by thin layer chromatographic (TLC) technique.

Course Outcomes: The experiments will make the student gain skills on:

- Determination of parameters like hardness and chloride content in water.
- Estimation of rate constant of a reaction from concentration time relationships.
- Determination of physical properties like adsorption and viscosity.
- Calculation of R_f values of some organic molecules by TLC technique.

List of Experiments:

- 1. Determination of total hardness of water by complexometric method using EDTA
- 2. Determination of chloride content of water by Argentometry
- 3. Estimation of an HCl by Conductometric titrations
- 4. Estimation of Acetic acid by Conductometric titrations
- 5. Estimation of HCl by Potentiometric titrations
- 6. Estimation of Fe²⁺ by Potentiometry using KMnO₄
- 7. Determination of rate constant of acid catalysed hydrolysis of methyl acetate
- 8. Synthesis of Aspirin and Paracetamol
- 9. Thin layer chromatography calculation of R_f values. eg ortho and para nitro phenols
- 10. Determination of acid value of coconut oil
- 11. Verification of freundlich adsorption isotherm-adsorption of acetic acid on charcoal
- 12. Determination of viscosity of castor oil and ground nut oil by using Ostwald's viscometer.
- 13. Determination of partition coefficient of acetic acid between n-butanol and water.
- 14. Determination of surface tension of a give liquid using stalagmometer.

References

- 1. Senior practical physical chemistry, B.D. Khosla, A. Gulati and V. Garg (R. Chand & Co., Delhi)
- 2. An introduction to practical chemistry, K.K. Sharma and D. S. Sharma (Vikas publishing, N. Delhi)
- 3. Vogel's text book of practical organic chemistry 5th edition
- 4. Text book on Experiments and calculations in Engineering chemistry S.S. Dara